

# Soluble Sulfate in the Martian Soil at the Phoenix Landing Site

Samuel P. Kounaves,<sup>1</sup> Michael H. Hecht,<sup>2</sup> Jason Kapit,<sup>1,3</sup> Richard C. Quinn,<sup>4</sup>  
David C. Catling,<sup>5</sup> Benton C. Clark,<sup>6</sup> Douglas W. Ming,<sup>7</sup> Kalina Gospodinova,<sup>1,8</sup>  
Patricia Hredzak,<sup>1</sup> Kyle McElhoney,<sup>1</sup> and Jennifer Shusterman<sup>1</sup>

<sup>1</sup> Department of Chemistry, Tufts University, Medford, Massachusetts, USA.

<sup>2</sup> Jet Propulsion Laboratory, California Institute of Technology, Pasadena, California, USA.

<sup>3</sup> Now at Woods Hole Oceanographic Institution, Woods Hole, Massachusetts, USA.

<sup>4</sup> SETI Institute, Moffett Field, California, USA.

<sup>5</sup> Department of Earth and Space Sciences, University of Washington, Seattle, Washington, USA.

<sup>6</sup> Space Science Institute, Boulder, Colorado, USA.

<sup>7</sup> NASA Johnson Space Center, Houston, Texas, USA.

<sup>8</sup> Now at Massachusetts Institute of Technology, Cambridge, Massachusetts, USA

Received 19 January 2010; revised 18 March 2010; accepted 29 March 2010; published 1 May 2010.

Sulfur has been detected by X-ray spectroscopy in martian soils at the Viking, Pathfinder, Opportunity and Spirit landing sites. Sulfates have been identified by OMEGA and CRISM in Valles Marineris and by the spectrometers on the MER rovers at Meridiani and Gusev. The ubiquitous presence of sulfur has been interpreted as a widely distributed sulfate mineralogy. One goal of the Wet Chemistry Laboratory (WCL) on NASA's Phoenix Mars Lander was to determine soluble sulfate in the martian soil. We report here the first in-situ measurement of soluble sulfate equivalent to  $\sim 1.3(\pm 0.5)$  wt% as  $\text{SO}_4$  in the soil. The results and models reveal  $\text{SO}_4^{2-}$  predominately as  $\text{MgSO}_4$  with some  $\text{CaSO}_4$ . If the soil had been wet in the past, epsomite and gypsum would be formed from evaporation. The WCL-derived salt composition indicates that if the soil at the Phoenix site were to form an aqueous solution by natural means, the water activity for a dilution of greater than  $\sim 0.015$  g  $\text{H}_2\text{O}/\text{g}$  soil would be in the habitable range of known terrestrial halophilic microbes.

Citation: Kounaves, S. P., *et al.* (2010), Soluble sulfate in the martian soil at the Phoenix landing site, *Geophys. Res. Lett.*, 37, L09201, doi:10.1029/2010GL042613.